

## Triterpenoid Saponins from *Schefflera divaricata*

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Received August 12, 1996<sup>®</sup>

Ten new triterpenoid saponins (**3–12**), along with two known triterpenoid saponins (**1**, **2**), were isolated from the aerial parts of *Schefflera divaricata* Merrill (Araliaceae). Their structures were determined by <sup>1</sup>H–<sup>1</sup>H correlation spectroscopy (COSY, HOHAHA) and <sup>1</sup>H–<sup>13</sup>C heteronuclear correlation (HETCOR, COLOC) NMR experiments, FABMS, and chemical data. The aglycon moieties are oleanolic acid for **1–4** and betulinic acid for **5–12**, respectively.

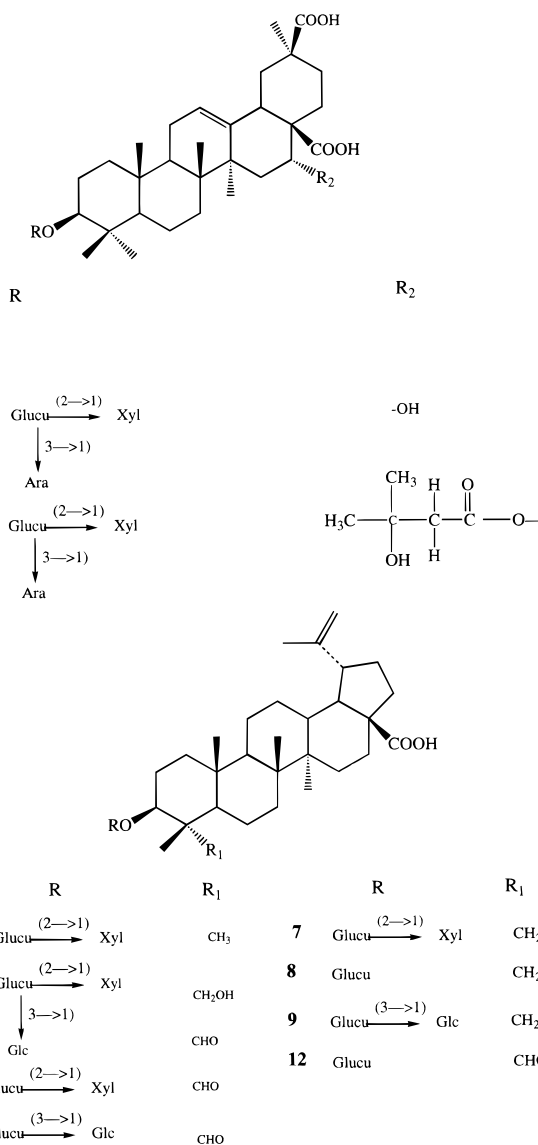
The Araliaceae family is one of the most medicinally important plant families. Perhaps the most famous member of this family is ginseng (*Panax ginseng* C.A. Meyer), the roots of which have been used in traditional oriental medicine for more than 5000 years.<sup>1</sup> Chemical and pharmacological investigations have indicated that triterpenoid saponins are important bioactive components existing in plants of the Araliaceae family.<sup>2</sup> Saponins isolated and identified from Araliaceae are reported to have various pharmacological activities including increasing mental efficiency, recovering physical balance, stimulation of metabolic function, and other general health-promoting effects.<sup>3</sup>

*Schefflera divaricata* Merrill (Araliaceae) is a plant native of Java and is used as a folk medicine for rheumatism and as a general tonic. No chemical work, however, has been reported on this plant. In a search for novel bioactive natural products from medicinal plants, we have studied the polar extract obtained from both flowers and leaves of *S. divaricata*. The present paper deals with the isolation and structure determination of oleanolic acids **3** and **4** and betulinic acid-based saponins **5–12** from the aerial parts of *S. divaricata*.

### Results and Discussion

The dried leaves of *Schefflera divaricata* were extracted successively with petroleum ether and MeOH. The MeOH extract was fractionated in H<sub>2</sub>O–*n*-BuOH, and the *n*-BuOH extract was subjected to Sephadex LH-20 column chromatography followed by DCCC to give saponins **1–12**. **1** and **2** were known compounds identified as (**1**) 3β-*O*-β-D-galactopyranosyl-(1→3)-[β-D-glucopyranosyl-(1→4)]-β-D-glucopyranosylolean-12-en-28-oic acid and (**2**) 3β-*O*-β-D-galactopyranosyl-(1→3)-[β-D-glucopyranosyl-(1→4)]-β-D-glucuronopyranosylolean-12-en-28-oic acid by spectral data and direct comparison of their physical properties with those reported previously for these compounds.<sup>4</sup>

The molecular formulas for all compounds were determined by <sup>13</sup>C, <sup>13</sup>C-DEPT NMR data, and FABMS in negative ion mode. <sup>1</sup>H and <sup>13</sup>C NMR spectra indicated that saponins **3** and **4** had identical saccharide chains but differed in the aglycon portion. The FABMS



of **3** (C<sub>46</sub>H<sub>72</sub>O<sub>19</sub>) showed the [M – H]<sup>–</sup> ion at *m/z* 927 with prominent fragments at *m/z* 795 [(M – H) – 132]<sup>–</sup> and 779 [(M – H) – 148]<sup>–</sup> (cleavage of a pentose unit with or without the glycosidic oxygen) and at *m/z* 485 [(M – H) – (132 + 176 + 132)]<sup>–</sup> due to the subsequent loss of two pentose and a hexose units. Also observed was a fragment at *m/z* 883 [(M – H) – 44]<sup>–</sup> corresponding to loss of a carboxylic group from the *m/z* 927

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<sup>®</sup> Abstract published in *Advance ACS Abstracts*, May, 1, 1997.

peak. The  $^{13}\text{C}$  and  $^{13}\text{C}$ -DEPT NMR spectra of **3** showed 46 signals of which 16 were assigned to the saccharide portion and 30 to a triterpene moiety. The  $^1\text{H}$  NMR spectrum of **3** showed six singlets assignable to tertiary methyl groups in the range  $\delta$  0.85–1.35. The  $3\beta$ -OH substitution was evident from the chemical shift and the  $J$  value of the proton assigned to C-3 centered at  $\delta$  3.20 (1H, dd,  $J = 11, 5.0$  Hz). The signal at  $\delta$  3.95 (br s) indicated the presence of a C-16 $\alpha$  hydroxyl group, a conclusion that was further supported by  $^{13}\text{C}$  NMR data (see the Experimental Section) and by the C-27 methyl, which resonated at  $\delta$  1.35, downfield from its usual position.<sup>5</sup> The  $^{13}\text{C}$  NMR of **3** also showed a quaternary signal at  $\delta$  180.00, indicative of a carboxylic group, which was assigned to C-30 on the basis of upfield shifts exhibited by C-19, C-21, and C-29 and the downfield shift experienced by C-20.<sup>6,7</sup> Thus, the aglycon of **3** was  $3\beta,16\alpha$ -dihydroxyolean-12-ene-28,30-dioic acid.

$^{13}\text{C}$  NMR signals assigned to the pentacyclic nucleus of the aglycon of compound **4** (see the Experimental Section) were similar to those reported for **3**. The FABMS of **4** ( $\text{C}_{51}\text{H}_{80}\text{O}_{21}$ ) showed an  $[\text{M} - \text{H}]^-$  ion at  $m/z$  1027 that was 100 mass units higher than that of **3** and was compatible with an additional 3-hydroxy-3-methylbutanoyl moiety. Analysis of its NMR spectral data indicated that **4** possessed the same basic structure of **3** but that it was esterified at C-16, with a 3-hydroxy-3-methylbutanoyl group. The  $^1\text{H}$  NMR spectrum was very similar to that of **3** with the main differences being a downfield shift of H-16 ( $\delta$  4.80 in **4** versus 3.95 in **3**), a signal at  $\delta$  2.25 (2H, s) for one methylene group, and a signal at  $\delta$  1.37 (6H, s) assigned to additional methyl groups. The  $^{13}\text{C}$  NMR of **4** showed five additional signals characteristic of a 3-hydroxy-3-methylbutanoyl moiety, and differences were observed in the chemical shifts of C-16 ( $\delta$  74.20 in **3** versus  $\delta$  76.00 in **4**) and C-15 ( $\delta$  35.20 in **3** versus  $\delta$  33.00 in **4**) that indicated that **4** was esterified at C-16.<sup>8</sup>

Attachment of the glycosidic chain at C-3 was indicated by the significant downfield shift ( $\delta$  90.60) observed for this carbon in **3** and **4** relative to the corresponding signal in similar compounds,<sup>4</sup> e.g., oleanolic acid ( $\delta$  78.00), and was subsequently confirmed by 2D-NMR experiments.<sup>9</sup> Structural elucidation of the glycosidic portion of these compounds began with the most abundant **3**. Acid methanolysis of the saponins gave glucuronic acid, xylose, and arabinose in a 1:1:1 ratio. This indicated that the sugar moiety linked to C-3 in **3** and **4** was a trisaccharide formed by glucuronic acid, xylose, and arabinose. The identity and sequence of sugar units in saponins **3** and **4** was deduced by a combination of 2D- $^1\text{H}$ - $^1\text{H}$  correlation spectroscopy COSY, HOHAHA,<sup>10</sup> and 2D  $^1\text{H}$ - $^{13}\text{C}$  correlation  $J^1$  (HETCOR and  $J$  long-range COLOC)<sup>11</sup> (Table 1). COSY and HOHAHA permitted assignments of all spin correlations from H-1 to H-5 of the glucuronic acid unit and H-1 to H-5 of the xylose and arabinose units. These data, together with results obtained from HETCOR and literature data,<sup>12,13</sup> established that the xylose and arabinose units were terminal and bonded to C-2 and C-3 of glucuronic acid (the sugar directly linked to the aglycon). These data left two possible sequences for the triglycoside chains of compounds **3** and **4**: a COLOC<sup>11</sup> experiment differentiated the two structures. A key correlation peak was obtained between the anomeric

**Table 1.**  $^1\text{H}$  NMR Data ( $J$  in Hz) for the Oligosaccharide Moieties of Compound **3** in  $\text{CD}_3\text{OD}^a$

position	<b>3</b>	
	$\delta\text{C}$	$\delta\text{H}$
Glucu 1	105.60	4.51 d, $J = 7.5$
2	82.90	3.35 s.o.
3	83.50	3.42 t, $J = 9.0$
4	70.20	3.50 t, $J = 9.0$
5	76.50	3.68 d, $J = 9.0$
6	172.00	
Xyl 1	105.30	4.49 d, $J = 7.5$
	75.50	3.20 dd, $J = 7.5, 9.5$
3	77.60	3.30 t, $J = 9.5$
4	72.80	3.52 m
5	67.10	3.90 dd, $J = 10.0, 5.0$
		3.18 dd, $J = 10.0, 2.0$
Ara 1	105.20	4.37 d, $J = 7.0$
2	73.90	3.80 dd, $J = 7.0, 8.5$
3	75.00	3.84 dd, $J = 8.5, 3.0$
4	69.60	4.00 m
5	65.20	4.02 dd, $J = 12.0, 2.0$
		3.63 dd, $J = 12.0, 3.5$

<sup>a</sup> Assignments were confirmed by  $^1\text{H}$ - $^1\text{H}$  COSY and HETCOR experiments.

proton of xylose ( $\delta$  4.49, d,  $J = 7.5$  Hz) and C-2 ( $\delta$  82.90) of the glucuronic acid unit. Chemical shift, multiplicity, absolute values of the coupling constant, and magnitude in the  $^1\text{H}$  NMR spectrum as well as  $^{13}\text{C}$  NMR data indicated the  $\beta$ -configuration at the anomeric positions for the xylose and glucose units and the  $\alpha$ -configuration for arabinose.<sup>14</sup>

Thus, the structures of **3** and **4** were determined to be  $3\beta$ - $O$ - $\alpha$ -L-arabinopyranosyl-(1 $\rightarrow$ 3)- $[\beta$ -D-xylopyranosyl-(1 $\rightarrow$ 2)]- $\beta$ -D-glucuronopyranosyl-16 $\alpha$ -hydroxyolean-12-ene-28,30-dioic acid **3**;  $3\beta$ - $O$ - $\alpha$ -L-arabinopyranosyl-(1 $\rightarrow$ 3)- $[\beta$ -D-xylopyranosyl-(1 $\rightarrow$ 2)]- $\beta$ -D-glucuronopyranosyl-16- $O$ -[3-hydroxy-3-methylbutanoyl]olean-12-ene-28,30-dioic acid **4**.

Compound **6** had the molecular formula  $\text{C}_{47}\text{H}_{74}\text{O}_{19}$ . The FABMS of **6** showed the  $[\text{M} - \text{H}]^-$  ion at  $m/z$  941 and fragments at  $m/z$  779  $[(\text{M} - \text{H}) - 162]^-$ ,  $m/z$  763  $[(\text{M} - \text{H}) - 178]^-$  (cleavage of a hexose unit with or without the glycosidic oxygen), and  $m/z$  647  $[(\text{M} - \text{H}) - (162 + 132)]^-$  due to the loss of a pentose unit, and  $m/z$  471  $[(\text{M} - \text{H}) - (132 + 162 + 176)]^-$ . The  $^{13}\text{C}$  and DEPT  $^{13}\text{C}$  NMR spectra showed 47 signals of which 17 were assigned to the saccharide portion and 30 to the triterpene moiety.

The  $^1\text{H}$  NMR spectrum of **6** showed, in addition to four methyl singlets assignable to tertiary methyls between  $\delta$  0.70–1.05, signals for an isopropylene function ( $\delta$  4.63 and 4.74, 1H each, br s and 1.73 3H, s) and signals for geminal protons at  $\delta$  3.30 and 3.70 (1H each, d,  $J = 12$  Hz,  $\text{H}_2$ -23) for the triterpene skeleton. A  $3\beta$ -OH substitution was evident from the chemical shift and the  $J$  values of the C-3 proton at  $\delta$  3.21 (1H, dd,  $J = 10.5, 4.0$  Hz). The  $^{13}\text{C}$  NMR spectrum of **6** suggested a lupene-type triterpene skeleton.<sup>15</sup> A signal at  $\delta$  64.70 (C-23) ( $\text{CH}_2$  by DEPT) and  $\delta$  13.20 (C-24) ( $\text{CH}_3$  by  $^{13}\text{C}$ -DEPT) indicated a  $-\text{CH}_2\text{OH}$  group at C-4. Glycosidation of the alcoholic function at C-3 was indicated by the significant downfield shift observed for this carbon in **6** (see the Experimental Section) relative to the corresponding signals in model compounds.<sup>15</sup>

The oligosaccharide structure was determined by 2D-NMR. Even at high field (500 MHz), the 1D sugar spectral region of **6** was complex as most of the shifts

**Table 2.**  $^{13}\text{C}$  NMR Data for the Oligosaccharide Moieties of Compounds **5**, **6**, **8**, and **9** in  $\text{CD}_3\text{OD}^a$ 

	<b>5</b> $\delta\text{C}$	<b>6</b> $\delta\text{C}$	<b>8</b> $\delta\text{C}$	<b>9</b> $\delta\text{C}$
Glucur 1	104.8	104.9	105.3	105.1
2	83.0	81.5	75.1	83.4
3	77.8	85.6	78.1	78.0
4	71.1	70.2	73.6	71.1
5	78.1	78.3	78.1	78.3
6	177.5	177.5	177.9	177.3
Xyl 1	106.3	106.0		
2	76.2	76.2		
3	78.1	78.0		
4	72.4	72.2		
5	67.1	66.5		
Glc 1		106.2		106.0
2		75.3		75.4
3		78.5		78.5
4		71.7		71.7
5		78.7		78.7
6		62.8		62.7

<sup>a</sup> Assignments were confirmed by  $^1\text{H}$ - $^1\text{H}$  COSY and HETCOR experiments.

were found between  $\delta$  5.00 and 3.00 and were overlapped by the aglycon signals. 2D-HOHAHA spectroscopy experiments allowed resolution of the overlapped spectra of oligosaccharides into a subset of individual monosaccharide spectra. In the 2D-HOHAHA spectrum of **6** the anomeric proton signal ascribable to an  $\beta$ -D-glucuronopyranose (H-1',  $\delta$  4.45,  $J = 7.0$  Hz) showed connectivities to four methines ( $\delta$  3.68, 3.50, 3.42, 3.35). This together with the 2D-COSY-90 spectrum established the proton sequence within this sugar fragment as H-1 ( $\delta$  4.45), H-2 ( $\delta$  3.35), H-3 ( $\delta$  3.42), H-4 ( $\delta$  3.50), and H-5 ( $\delta$  3.68) (Table 3).

Similar observations of the HOHAHA and COSY experiments for all the other sugar residues (Tables 2 and 3) allowed complete sequential assignments for all proton resonances starting from the anomeric proton signals. HETCOR experiments that correlated all proton resonances with those of each corresponding carbon (Tables 2 and 3) permitted assignments of the interglycosidic linkages by comparison of the  $^{13}\text{C}$  shifts observed with those of the corresponding methyl pyranosides and taking into account the known effects of glycosidation.<sup>16</sup> The absence of any  $^{13}\text{C}$  glycosidation shift for xylopyranosyl and glucopyranosyl residues suggested that these sugars were terminal units, while the glycosidation shifts on C-2 ( $\sim +7$  ppm) and C-3 ( $\sim +7$ ) of the glucuronopyranosyl unit indicated the presence of a glucuronopyranosyl residue glycosylated at C-2 and C-3. The position of each sugar unit was deduced from a 2D-ROESY<sup>14</sup> experiment that showed a cross peak between the signal at  $\delta$  4.45 (H-1 of the glucuronic acid) and the signal at  $\delta$  3.20 (H-3 of the aglycone moiety) and other key correlation peaks between the signal at  $\delta$  4.66 (H-1 of glucose) and the signal at  $\delta$  3.42 (H-3 of the glucuronic acid moiety). The  $^1\text{H}$  NMR and  $^{13}\text{C}$  NMR data indicated the  $\beta$  configuration at the anomeric positions for all sugar units.<sup>14</sup> Therefore, the structure  $3\beta$ - $O$ - $\beta$ -D-glucopyranosyl-(1 $\rightarrow$ 3)-[ $\beta$ -D-xylopyranosyl-(1 $\rightarrow$ 2)]- $\beta$ -D-glucuronopyranosyl-23-hydroxylup-20(29)-en-28-oic acid was assigned to **6**.

Compound **7** had the molecular formula  $\text{C}_{41}\text{H}_{64}\text{O}_{14}$ . The FABMS spectrum of **7** showed the  $[\text{M} - \text{H}]^-$  ion at  $m/z$  779 and prominent fragments at  $m/z$  647  $[(\text{M} - \text{H}) - 132]^-$ ;  $m/z$  631  $[(\text{M} - \text{H}) - 148]^-$  (cleavage of a pentose unit with or without the glycosidic oxygen); and

$m/z$  471  $[(\text{M} - \text{H}) - (132 + 176)]^-$  due to the subsequent loss of an uronic acid unit. The  $^{13}\text{C}$  and DEPT  $^{13}\text{C}$  NMR spectra showed 41 signals of which 11 were assigned to the saccharide portion and 30 to a triterpenic moiety. Analysis of NMR data of compound **7** and comparison with those of **6** showed **7** to differ from **6** only in the absence of the terminal glucopyranosyl unit (Tables 2 and 3). Therefore, the structure  $3\beta$ - $O$ - $\beta$ -D-xylopyranosyl-(1 $\rightarrow$ 2)- $\beta$ -D-glucuronopyranosyl-23-hydroxylup-20(29)-en-28-oic acid was assigned to **7**.

The FABMS spectrum of **8** ( $\text{C}_{36}\text{H}_{56}\text{O}_{10}$ ) showed the  $[\text{M} - \text{H}]^-$  ion at  $m/z$  647 and prominent fragments at  $m/z$  471  $[(\text{M} - \text{H}) - 176]^-$  due to the loss of a glucuronic acid unit. The NMR spectra of compound **8** exhibited all resonances of an unsubstituted  $\beta$ -D-glucuronopyranosyl residue and of the same pentacyclic ring system as compound **6** (Tables 2 and 3). Thus, compound **8** was determined to be  $3\beta$ - $O$ - $\beta$ -D-glucuronopyranosyl-23-hydroxylup-20(29)-en-28-oic acid.

Compound **9** had the molecular formula  $\text{C}_{42}\text{H}_{66}\text{O}_{15}$ . The FABMS spectrum of **9** showed the  $[\text{M} - \text{H}]^-$  ion at  $m/z$  809 and prominent fragments at  $m/z$  631  $[(\text{M} - \text{H}) - 178]^-$ ,  $m/z$  647  $[(\text{M} - \text{H}) - 162]^-$  (cleavage of a hexose unit with or without the glycosidic oxygen), and  $m/z$  471  $[(\text{M} - \text{H}) - (162 + 176)]^-$  due to the loss of a glucuronic acid unit. The  $^{13}\text{C}$  and DEPT  $^{13}\text{C}$  NMR spectra showed 42 signals of which 12 were assigned to the saccharide portion and 30 to a triterpene moiety. The analysis of NMR data of compound **9** and comparison with those of **6** showed **9** to differ from **6** only for the absence of the terminal xylopyranosyl unit (Tables 2 and 3). Thus, compound **9** was determined to be  $3\beta$ - $O$ - $\beta$ -D-glucopyranosyl-(1 $\rightarrow$ 3)- $\beta$ -D-glucuronopyranosyl-23-hydroxylup-20(29)-en-28-oic acid.

Compound **5** ( $\text{C}_{41}\text{H}_{64}\text{O}_{13}$ ) exhibited a quasimolecular ion at  $m/z$  763 in its FAB-MS spectrum, as well as signals at  $m/z$  631  $[(\text{M} - \text{H}) - 132]^-$  and  $m/z$  455  $[(\text{M} - \text{H}) - (132 + 176)]^-$  corresponding to the sequential losses of pentose and uronic acid units. The  $^{13}\text{C}$  NMR spectrum showed 41 carbon signals of which 11 were assigned to the saccharide portion and 30 to a triterpene moiety. Acid hydrolysis of **5** yielded betulinic acid,<sup>18,19</sup> xylose, and glucuronic acid (GC) in the molar ratio 1:1:1. The spectral data for the aglycon moiety of **5** were identical with those of betulinic acid.<sup>18,19</sup> Comparison of NMR data of **5** with those of **7** indicated that both had the same glycosidic unit. Thus, compound **5** was determined to be  $3\beta$ - $O$ - $\beta$ -D-xylopyranosyl-(1 $\rightarrow$ 2)- $\beta$ -D-glucuronopyranosylup-20(29)-en-28-oic acid.

Compound **10** had the molecular formula  $\text{C}_{41}\text{H}_{64}\text{O}_{13}$ . Comparison of the NMR spectral data of **10** with those of **5** indicated that **10** was a  $3\beta$ -hydroxylup-20(29)-en-28-oic acid derivative. In particular, hydrogens and carbons due to the C, D, E ring of **10** resonated near the same frequencies as the corresponding signals in betulinic acid, while the A- and B-ring  $^1\text{H}$  and  $^{13}\text{C}$  signals were shifted somewhat. The NMR spectra of **10** contained one less methyl and one more signal [ $^1\text{H}$  NMR  $\delta$  9.50 (1H, s);  $^{13}\text{C}$  NMR  $\delta$  209.10 (CH)] than those of **5**, suggesting that one of the Me groups was replaced by a formyl group in **10**. The  $^1\text{H}$  spectrum of **10** showed that the H-3 ( $\delta$  3.66) signals were shifted downfield by 0.5 ppm in comparison with those of **5**. In addition, one of the methyl signals was shifted downfield to  $\delta$  1.40 in **10** due to the formyl group. Therefore, the  $-\text{CHO}$  group

**Table 3.**  $^1\text{H}$  NMR Data ( $J$  in Hz) for the Oligosaccharide Moieties of Compounds **5**, **6**, **8**, and **9** in  $\text{CD}_3\text{OD}^a$ 

	<b>5</b> $\delta\text{H}$	<b>6</b> $\delta\text{H}$	<b>8</b> $\delta\text{H}$	<b>9</b> $\delta\text{H}$
Glucur 1	4.47 d, $J = 7.0$	4.45 d, $J = 7.0$	4.48 d, $J = 7.0$	4.48 d, $J = 7.0$
2	3.35 dd, $J = 7.0, 9.0$	3.35 dd, $J = 7.0, 9.0$	3.32 dd, $J = 7.0, 9.0$	3.33 dd, $J = 7.0, 9.0$
3	3.40 t, $J = 9.0$	3.42 t, $J = 9.0$	3.40 t, $J = 9.0$	3.40 t, $J = 9.0$
4	3.45 dd, $J = 9.0, 10.0$	3.50 dd, $J = 9.0, 10.0$	3.47 dd, $J = 9.0, 10.0$	3.48 dd, $J = 9.0, 10.0$
5	3.65 d, $J = 10.0$	3.68 d, $J = 10.0$	3.63 d, $J = 10.0$	3.64 d, $J = 10.0$
Xyl 1	4.55 d, $J = 7.6$	4.52 d, $J = 7.5$		
2	3.20 dd, $J = 7.6, 9.0$	3.24 dd, $J = 7.5, 9.0$		
3	3.30 t, $J = 9.0$	3.45 $J = 9.0$		
4	3.50 ddd, $J = 10.0, 9.0, 5.0$	3.53 ddd, $J = 10.0, 9.0, 5.0$		
5a	3.89 dd, $J = 10.0, 12.0$	3.86 dd, $J = 10.0, 11.0$		
5b	3.12 dd, $J = 5.0, 12.0$	3.20 dd, $J = 5.0, 11.0$		
Glc 1		4.66 d, $J = 7.5$		4.62 d, $J = 7.5$
2		3.28 dd, $J = 9.0, 7.5$		3.28 dd, $J = 9.0, 7.5$
3		3.32 t, $J = 9.0$		3.31 t, $J = 9.0$
4		3.37 t, $J = 9.0$		3.38 t, $J = 9.0$
5		3.40 m		3.34 m
6a		3.85 dd, $J = 12.0, 2.5$		3.81 dd, $J = 12.0, 2.5$
6b		3.64 dd, $J = 12.0, 6.0$		3.62 dd, $J = 12.0, 6.0$

<sup>a</sup> Assignments are confirmed by  $^1\text{H}$ - $^1\text{H}$  COSY, HOHAHA, and HETCOR experiments.

could only be at C-23 or C-24.<sup>20</sup> The most significant features of the  $^{13}\text{C}$  NMR spectrum of **10**, which suggested placement of the formyl group at C-23, were the downfield shifts exhibited by C-4 and C-6 and the upfield shifts experienced by C-3, C-5, and Me-24. The aglycon of **10** was elucidated to be  $3\beta$ -hydroxy-23-oxolup-20(29)-en-28-oic acid and has not been reported previously. Analysis of NMR data of **10** for resonances of the glycosidic moiety indicated that **10** had the same glycosidic chain as **7**. Thus, **10** was determined to be  $3\beta$ - $O$ - $\beta$ -D-xylopyranosyl-(1 $\rightarrow$ 2)- $\beta$ -D-glucuronopyranosyl-23-oxolup-20(29)-en-28-oic acid.

The FABMS of compound **11** ( $\text{C}_{42}\text{H}_{64}\text{O}_{15}$ ) displayed a molecular ion peak at  $m/z$  807  $[\text{M} - \text{H}]^-$  and prominent fragments at  $m/z$  645  $[(\text{M} - \text{H}) - 162]^-$  and  $m/z$  469  $[(\text{M} - \text{H}) - (162 + 176)]^-$  due to the subsequent losses of a hexose and glucuronic acid units. The  $^{13}\text{C}$  aglycon signals were similar to those of **10** allowing identification of the aglycon of **11** as  $3\beta$ -hydroxy-23-oxolup-20(29)-en-28-oic acid. Analysis of the spectral data for **11** revealed that this compound had the same glycosidic chain as compound **9**. Therefore, compound **11** was determined to be  $3\beta$ - $O$ - $\beta$ -D-glucopyranosyl-(1 $\rightarrow$ 3)- $\beta$ -D-glucuronopyranosyl-23-oxolup-20(29)-en-28-oic acid.

The FABMS spectrum of compound **12** ( $\text{C}_{36}\text{H}_{54}\text{O}_{10}$ ) displayed a molecular ion peak at  $m/z$  645  $[\text{M} - \text{H}]^-$  and a fragment at  $m/z$  469  $[(\text{M} - \text{H}) - 176]^-$  corresponding to the loss of a glucuronic acid residue. The NMR spectra of compound **12** exhibited all resonances of unsubstituted  $\beta$ -D-glucuronopyranose and the same pentacyclic ring system as compound **10**. Thus, **12** was concluded to be  $3\beta$ - $O$ - $\beta$ -D-glucuronopyranosyl-23-oxolup-20(29)-en-28-oic acid.

## Experimental Section

**General Experimental Procedures.** A Bruker AMX-500 spectrometer equipped with a Bruker X-32 computer using the UxNMR software package was used for NMR measurements in  $\text{CD}_3\text{OD}$  solutions. 2D-homonuclear proton chemical shift correlation (COSY), 2D-HOHAHA, ROESY,  $^1\text{H}$ - $^{13}\text{C}$  HETCOR, and COLOC experiments were obtained as described previously.<sup>10,11</sup> Optical rotations were measured on a Perkin-Elmer 141 polarimeter using a sodium lamp operating at 589 nm in 1% w/v solutions in MeOH. Fast atom bombardment

mass spectra (FABMS) were recorded in a glycerol matrix in the negative ion mode on a VG ZAB instrument (XE atoms of energy of 2–6 KV). DCCC separations were performed on a Buchi apparatus, equipped with 300 tubes. HPLC separations were performed using a Waters Model 6000A pump equipped with a U6K injector and a Model 401 refractive index detector.

**Plant Material.** The plant *S. divaricata* (Araliaceae) was collected in Palermo, Italy, in the spring of 1994; a sample has been deposited in the Herbarium of the Botanical Garden of Palermo.

**Extraction and Isolation.** Dried flowers of *S. divaricata* (500 g) were defatted with petroleum ether then extracted with MeOH to give (50 g) of residue. The methanolic extract was dissolved in  $\text{H}_2\text{O}$ . The  $\text{H}_2\text{O}$  extract (40 g) was partitioned between AcOEt and *n*-BuOH to afford an *n*-BuOH-soluble portion (6 g) and an AcOEt portion (1 g). The *n*-BuOH extract was chromatographed on a Sephadex LH-20 column (100  $\times$  5 cm) with MeOH as the eluent. Fractions (8 mL) were collected and checked by TLC. Fractions 16–23 (540 mg) from Sephadex containing a mixture of oleanolic glycosides **1** and **2** were separated by RPHPLC on a C18  $\mu$ -Bondapak column (30 cm  $\times$  7.8 mm, flow rate 2.0 mL/min) with MeOH- $\text{H}_2\text{O}$  (1:1) to yield pure compounds **1** ( $t_R = 25$  min, 22.5 mg), **2** ( $t_R = 17$  min, 16 mg). The AcOEt residue (1 g) contained pure compound **1**.

The air-dried leaves of *S. divaricata* (300 g) were defatted with petroleum ether and then extracted with MeOH to give (27 g) of residue. The methanolic extract was dissolved in  $\text{H}_2\text{O}$ . The  $\text{H}_2\text{O}$  extract (13 g) was partitioned between AcOEt and *n*-BuOH to afford an *n*-BuOH-soluble portion (3.6 g) that was chromatographed on a Sephadex LH-20 column (100  $\times$  5 cm) with MeOH as the eluent. Fractions (8 mL) were collected and checked by TLC.

Fractions 18–25 (540 mg) from Sephadex were purified by DCCC with BuOH-EtOH-AcOH- $\text{H}_2\text{O}$  (8:4:2:1) in which the stationary phase consisted of the higher phase (ascending mode, flow 10 mL/h). About 250 fractions (4 mL) were collected. DCCC fractions 60–158 (250 mg) containing a mixture of oleanolic glycosides **1**–**4** were separated by RP-HPLC on a C18  $\mu$ -Bondapak column (30 cm  $\times$  7.8 mm, flow rate 2.0 mL/min) with MeOH- $\text{H}_2\text{O}$  (1:1) to yield pure compounds **1**

( $t_R$  = 25 min, 22.5 mg), **2** ( $t_R$  = 17 min, 16 mg), **3** ( $t_R$  = 10 min, 28 mg), and **4** ( $t_R$  = 12.5 min, 21 mg).

Fractions 28–34 (1g) from Sephadex were purified by DCCC [CHCl<sub>3</sub>–H<sub>2</sub>O–EtOH–MeOH–*n*-PrOH (9:8:8:6:1), ascending mode, flow 10 mL/h]. DCCC fractions 110–145 (150 mg) and 160–195 (550 mg) containing a mixture of lupene glycosides were further separated by RP-HPLC on a C18  $\mu$ -Bondapak column (30 cm  $\times$  7.8 mm, flow rate 2.0 mL/min) with MeOH–H<sub>2</sub>O (7:3) to yield the following: from fractions 110–145, compounds **5** ( $t_R$  = 25 min, 40 mg), **8** ( $t_R$  = 17 min, 19 mg), **12** ( $t_R$  = 21 min, 12 mg); with MeOH–H<sub>2</sub>O (65:25): from fractions 160–195, compounds **6** ( $t_R$  = 8 min, 38 mg), **7** ( $t_R$  = 13 min, 28 mg), **9** ( $t_R$  = 11 min, 20 mg), **10** ( $t_R$  = 16 min, 18 mg), and **11** ( $t_R$  = 14.5 min, 12 mg).

**Methanolysis of Compounds 3–12, Carbohydrate Constituents.** A solution of compound (2 mg) in anhydrous 2 N HCl–MeOH (0.5 mL) was heated at 80 °C in a stoppered reaction vial for 12 h. After cooling, the solution was neutralized with Ag<sub>2</sub>CO<sub>3</sub> and centrifuged, and then the supernatant was evaporated to dryness under N<sub>2</sub>. The residue was reacted with TRISIL-Z (Pierce) and analyzed by GLC. Retention times were identical to those of authentic methyl sugars.

**Compound 1** was identified as 3-*O*-[ $\beta$ -D-galactopyranosyl-(1 $\rightarrow$ 3)]- $\beta$ -D-glucopyranosyl-(1 $\rightarrow$ 4)]- $\beta$ -D-glucopyranosyl]olean-12-en-28-oic acid, previously isolated from *Calendula arvensis*, by spectral data.<sup>4</sup>

**Compound 2** was identified as 3-*O*-[ $\beta$ -D-galactopyranosyl-(1 $\rightarrow$ 3)]- $\beta$ -D-glucopyranosyl-(1 $\rightarrow$ 4)]- $\beta$ -D-glucuronopyranosyl]olean-12-en-28-oic acid, previously isolated from *Calendula arvensis*, by spectral data.<sup>4</sup>

**Compound 3:**  $[\alpha]^{25}_D = +9.0^\circ$  (*c* 1, MeOH); C<sub>46</sub>H<sub>72</sub>O<sub>19</sub>, negative FABMS  $m/z$  [M – H]<sup>–</sup> 927, [(M – H) – 132]<sup>–</sup> 795, [(M – H) – 148]<sup>–</sup> 779, [(M – H) – (132 + 176 + 132)]<sup>–</sup> 485; <sup>1</sup>H NMR for aglycon (CD<sub>3</sub>OD, 500 MHz)  $\delta$  0.92 (3H, s, Me-24), 0.99 (3H, s, Me-25), 1.09 (3H, s, Me-23), 1.19 (3H, s, Me-26), 1.18 (3H, s, Me-29), 1.35 (3H, s, Me-27), 3.20 (1H, dd,  $J$  = 11.0, 5.0 Hz), 3.95 (1H, br s, H-16), 2.88 (1H, dd,  $J$  = 12.5, 5.0 Hz, H-18), 3.20 (1H, dd,  $J$  = 11.0, 5.0 Hz, H-3); <sup>13</sup>C NMR for aglycon (CD<sub>3</sub>OD, 500 MHz)  $\delta$  16.1 (C-24), 16.7 (C-25), 17.9 (C-26), 19.4 (C-6), 24.5 (C-11), 26.3 (C-2), 26.4 (C-27), 27.4 (C-23), 27.4 (C-21), 28.3 (C-29), 31.1 (C-7), 32.4 (C-22), 35.2 (C-15), 36.6 (C-20), 37.3 (C-10), 38.9 (C-8), 39.36 (C-4), 40.3 (C-14), 42.3 (C-18), 42.7 (C-19), 52.2 (C-17), 57.2 (C-5), 74.2 (C-16), 90.6 (C-3), 123.5 (C-12), 145.0 (C-13), 181.9 (C-28), 180.0 (C-30). For NMR data of the sugar moiety see Table 1.

**Compound 4:**  $[\alpha]^{25}_D = +3.4^\circ$  (*c* 1, MeOH); C<sub>51</sub>H<sub>80</sub>O<sub>21</sub>, negative FABMS  $m/z$  [M – H]<sup>–</sup> 1027, [(M – H) – 132]<sup>–</sup> 895; <sup>1</sup>H NMR for aglycon (CD<sub>3</sub>OD, 500 MHz)  $\delta$  0.90 (3H, s, Me-24), 0.99 (3H, s, Me-25), 1.12 (3H, s, Me-23), 1.15 (3H, s, Me-26), 1.16 (3H, s, Me-29), 1.36 (3H, s, Me-27), 1.47 (6H, s, Me-4' and Me-5') 2.25 (2H, br s, H-2'), 2.90 (1H, dd,  $J$  = 12.5, 5.0 Hz, H-18), 3.20 (1H, dd,  $J$  = 11.0, 5.0 Hz, H-3), 4.80 (1H, br s, H-16); <sup>13</sup>C NMR for aglycon (CD<sub>3</sub>OD, 500 MHz)  $\delta$  16.1 (C-24), 16.6 (C-25), 17.9 (C-26), 19.1 (C-6), 24.4 (C-11), 26.5 (C-2), 26.5 (C-27), 27.2 (C-23), 27.8 (C-21), 28.2 (C-29), 32.0 (C-7), 32.5 (C-22), 33.0 (C-15), 36.7 (C-20), 37.4 (C-10), 38.8 (C-8), 39.0 (C-4), 40.4 (C-14), 42.5 (C-18), 42.7 (C-19), 50.0 (C-17), 57.0 (C-5), 76.0 (C-16), 90.0 (C-3), 123.5 (C-12), 144.9 (C-13), 176.7 (C-28), 180.1 (C-30), 21.2 (C-4), 23.4 (C-5'), 70.2 (C-3'), 42.0 (C-2'), 175.0 (C-1'). NMR

data of the sugar moiety are identical to those reported for compound **3**.

**Compound 5:**  $[\alpha]^{25}_D +112^\circ$  (*c* 1, MeOH); C<sub>41</sub>H<sub>64</sub>O<sub>13</sub>, negative FABMS  $m/z$  [M – H]<sup>–</sup> 763, [(M – H) – 132]<sup>–</sup> 631, [(M – H) – 148]<sup>–</sup> 647, [(M – H) – (132 + 176)]<sup>–</sup> 455; <sup>1</sup>H NMR for aglycon (CD<sub>3</sub>OD, 500 MHz)  $\delta$  0.72 (1H, dd,  $J$  = 11.0, 2.5 Hz, H-5), 0.76 (3H, s, Me-24), 0.81 (3H, s, Me-25), 0.90 (3H, s, Me-26), 0.95 (3H, s, Me-23), 1.00 (3H, s, Me-27), 1.73 (1H, s, Me-30), 1.86 (1H, dd,  $J$  = 8.0, 11.5 Hz, H-18), 2.82 (1H, br ddd,  $J$  = 11.5, 11.5, 5.0 Hz, H-19), 3.20 (1H, dd,  $J$  = 11.5, 5.0 Hz, H-3), 4.62 (1H, br s, H-29), 4.74 (1H, br s, H-29); <sup>13</sup>C NMR for aglycon (CD<sub>3</sub>OD, 500 MHz)  $\delta$  15.1 (C-27), 16.0 (C-25), 16.4 (C-26), 16.0 (C-25), 16.5 (C-24), 18.1 (C-6), 20.0 (C-30), 21.9 (C-11), 26.5 (C-2), 26.9 (C-12), 28.0 (C-23), 28.2 (C-15), 30.2 (C-21), 33.1 (C-7), 32.7 (C-16), 37.4 (C-10), 37.9 (C-13), 38.0 (C-22), 40.2 (C-8), 39.9 (C-1), 39.0 (C-4), 42.8 (C-14), 47.4 (C-18), 57.0 (C-5), 51.0 (C-19), 50.3 (C-9), 56.3 (C-17), 90.0 (C-3), 111.0 (C-29), 151.8 (C-20), 182.0 (C-28). For NMR data of the sugar moiety see tables 2 and 3.

**Compound 6:**  $[\alpha]^{25}_D +95^\circ$  (*c* 1, MeOH); C<sub>47</sub>H<sub>74</sub>O<sub>19</sub>, negative FABMS  $m/z$  [M – H]<sup>–</sup> 941, [(M – H) – 162]<sup>–</sup> 779, [(M – H) – (162 + 132)]<sup>–</sup> 647, [(M – H) – (162 + 148)]<sup>–</sup> 663, [(M – H) – (162 + 132 + 176)]<sup>–</sup> 471; <sup>1</sup>H NMR for aglycon (CD<sub>3</sub>OD, 500 MHz)  $\delta$  0.71 (3H, s, Me-24), 0.91 (3H, s, Me-25), 1.05 (3H, s, Me-26), 1.00 (3H, s, Me-27), 1.73 (1H, s, Me-30), 1.86 (1H, dd,  $J$  = 8.0, 11.5 Hz, H-18), 2.26 (1H, br ddd,  $J$  = 11.5, 11.5, 5.0 Hz, H-19), 2.32 (1H, br ddd,  $J$  = 11.0, 11.5, 3.0 Hz, H-13), 3.21 (1H, dd,  $J$  = 10.5, 4.0 Hz, H-3), 3.30 (1H, d,  $J$  = 12.0 Hz, Ha-23), 3.70 (1H, d,  $J$  = 12.0 Hz, Hb-23), 4.63 (1H, br s, Ha-29), 4.74 (1H, br s, Hb-29); <sup>13</sup>C NMR for aglycon (CD<sub>3</sub>OD, 500 MHz)  $\delta$  13.20 (C-24), 15.1 (C-27), 16.1 (C-25), 16.2 (C-26), 18.9 (C-6), 19.60 (C-30), 22.2 (C-11), 26.4 (C-2), 26.9 (C-12), 28.6 (C-15), 30.1 (C-21), 31.1 (C-7), 32.9 (C-16), 37.1 (C-10), 38.0 (C-13), 38.2 (C-22), 39.7 (C-8), 39.8 (C-1), 43.7 (C-4), 44.0 (C-14), 48.1 (C-18), 48.4 (C-5), 50.2 (C-19), 52.0 (C-9), 57.5 (C-17), 64.7 (C-24), 82.4 (C-3), 110.3 (C-29), 150.3 (C-20), 181.7 (C-28). For <sup>13</sup>C NMR data of the sugar moiety see Tables 2 and 3.

**Compound 7:**  $[\alpha]^{25}_D +105^\circ$  (*c* 1, MeOH); C<sub>41</sub>H<sub>64</sub>O<sub>14</sub>, negative FABMS  $m/z$  [M – H]<sup>–</sup> 779, [(M – H) – 132]<sup>–</sup> 647, [(M – H) – (132 + 176)]<sup>–</sup> 471. NMR data for the aglycon moiety are identical to those for **6**; for the sugar moiety signals are superimposable on those reported for **5**.

**Compound 8:**  $[\alpha]^{25}_D +60^\circ$  (*c* 1, MeOH); C<sub>36</sub>H<sub>56</sub>O<sub>10</sub>, negative FABMS  $m/z$  [M – H]<sup>–</sup> 647, [(M – H) – 176]<sup>–</sup> 471. NMR data for the aglycon moiety are identical to those for compound **6**; for the sugar moiety see Tables 2 and 3.

**Compound 9:**  $[\alpha]^{25}_D +101^\circ$  (*c* 1, MeOH); C<sub>42</sub>H<sub>66</sub>O<sub>15</sub>, negative FABMS  $m/z$  [M – H]<sup>–</sup> 809, [(M – H) – 162]<sup>–</sup> 647, [(M – H) – (162 + 176)]<sup>–</sup> 471. NMR data for the aglycon moiety are identical to those for **6**; for the sugar moiety see Tables 2 and 3.

**Compound 10:**  $[\alpha]^{25}_D +120^\circ$  (*c* 1, MeOH); C<sub>41</sub>H<sub>64</sub>O<sub>13</sub>, negative FABMS  $m/z$  [M – H]<sup>–</sup> 777, [(M – H) – 132]<sup>–</sup> 645, [(M – H) – (132 + 176)]<sup>–</sup> 469; <sup>1</sup>H NMR for aglycon (CD<sub>3</sub>OD, 500 MHz)  $\delta$  0.94 (3H, s, Me-25), 1.06 (3H, s, Me-26), 1.00 (3H, s, Me-27), 1.40 (3H, s, Me-24), 1.75 (1H, s, Me-30), 1.89 (1H, dd,  $J$  = 8.0, 11.5 Hz, H-18), 2.28 (1H, br ddd,  $J$  = 11.5, 11.5, 5.0 Hz, H-19), 2.33 (1H,

br ddd,  $J = 11.0, 11.5, 3.0$  Hz, H-13), 3.66 (1H, dd,  $J = 11.5, 5.0$  Hz, H-3), 4.63 (1H, br s, Ha-29), 4.74 (1H, br s, Hb-29), 9.50 (1H, s, H-23);  $^{13}\text{C}$  NMR for aglycon ( $\text{CD}_3\text{OD}$ , 500 MHz)  $\delta$  11.0 (C-24), 15.0 (C-27), 15.8 (C-25), 16.0 (C-26), 19.60 (C-30), 20.4 (C-6), 22.0 (C-11), 25.2 (C-2), 26.0 (C-12), 29.8 (C-21), 30.5 (C-15), 32.5 (C-7), 32.0 (C-16), 36.8 (C-10), 38.7 (C-13), 37.0 (C-22), 40.4 (C-8), 38.3 (C-1), 48.5 (C-5), 42.4 (C-14), 46.9 (C-18), 49.9 (C-19), 49.8 (C-9), 55.1 (C-4), 56.3 (C-17), 84.2 (C-3), 110.6 (C-29), 151.0 (C-20), 182.0 (C-28), 209.1 (C-23).  $^{13}\text{C}$  NMR data for the sugar moiety signals were superimposable on those reported for compound **5**.

**Compound 11:**  $[\alpha]_D^{25} +98^\circ$  ( $c$  1, MeOH);  $\text{C}_{42}\text{H}_{64}\text{O}_{15}$ , negative FABMS  $m/z$   $[\text{M} - \text{H}]^-$  807,  $[(\text{M} - \text{H}) - 162]^-$  645,  $[(\text{M} - \text{H}) - (162 + 176)]^-$  469. NMR data for the aglycon moiety are superimposable on those for **10**; for the sugar moiety see Tables 2 and 3.

**Compound 12:**  $[\alpha]_D^{25} +62$  ( $c$  1, MeOH);  $\text{C}_{36}\text{H}_{54}\text{O}_{10}$ , negative FABMS  $m/z$   $[\text{M} - \text{H}]^-$  645,  $[(\text{M} - \text{H}) - 176]^-$  469; NMR data for the aglycon moiety are identical to those for **10**; for the sugar moiety signals are superimposable on those reported for **8**.

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NP9605807